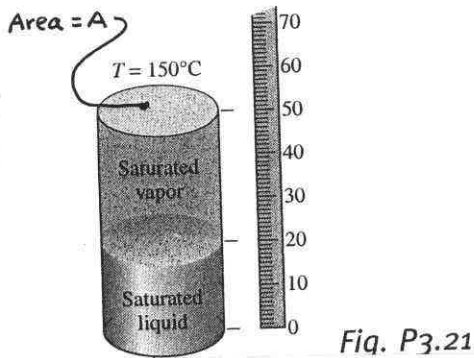


PROBLEM 3.21

KNOWN: A closed, rigid container holds different volumes of saturated liquid water and saturated vapor water.

FIND: Determine the quality of the mixture.

SCHEMATIC & GIVEN DATA:



ANALYSIS: $x = \frac{m_{\text{vap}}}{m_{\text{vap}} + m_{\text{liq}}}$, $m = V/v$. Thus, $m_{\text{vap}} = \frac{V_{\text{vap}}}{v_g}$, $m_{\text{liq}} = \frac{V_{\text{liq}}}{v_f}$

$$x = \frac{V_{\text{vap}}/v_g}{(V_{\text{vap}}/v_g) + (V_{\text{liq}}/v_f)}$$

$V_{\text{vap}} = 30A$ and $V_{\text{liq}} = 20A$, where area A is in the same units as the vertical measure shown. Then

$$x = \frac{(30A/v_g)}{(30A/v_g) + (20A/v_f)} = \frac{1}{1 + \frac{20}{30} \left(\frac{v_g}{v_f} \right)}$$

Since ratios appear in the last expression, the quantities can be in any consistent units.

① Using v_f and v_g from Table A-2 at 150°C ,

$$v_f = 1.0905 \times 10^{-3} \text{ m}^3/\text{kg}$$

$$v_g = 0.3928 \text{ m}^3/\text{kg}$$

$$x = \frac{1}{1 + \frac{20}{30} \left(\frac{0.3928}{1.0905 \times 10^{-3}} \right)} = 0.0041 \text{ (0.41\%)} \quad \leftarrow x$$

1. Using v_f and v_g at 302°F (150°C) from Table A-2E: $v_f = 0.017468 \text{ ft}^3/\text{lb}$, $v_g = 6.292 \text{ ft}^3/\text{lb}$ gives the same value for x , as can be verified.